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K, °C Section 1 Answers 3 and 4 are interchangeable. 3. observation Section 3 4. investigation 1. line graph 5. overlap 2. temperature vs. time for heating of water 6. explanations 3. time 7. technology 4. horizontal or x-axis 8. modified 5. vertical or y-axis Answers 9, 10, and 11 are interchangeable. 9.

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Study Guide and Reinforcement - Answer Key

Chapter 2 Review Answers. Section 1 Reinforcement - Describing Motion. 1. e. 2. b. 3. a and c. 4. d. 5. walking home. 6. You can tell an object has moved because its position has changed. 7. Displacement is how far and in what direction an object has moved from its starting point. 8.

Physics with Ms. Christenson - Home

Study Guide and Reinforcement 3 ANSWER KEY 7. opposes the motion of objects that move through the air, is affected by speed, size, and shape 8. net force 9. microwelds 10. rolling 11. air resistance 12. acceleration 13. sliding 14. parachute 1. Gravity is a force that every object in the

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[DOC] 3 1 Review Reinforcement Answer Key 10—2 Review and Reinforcement (continued) 9. $0.0621 \text{ mol SS} = (Q \text{ } \$ \text{ gK}2\text{S } 10. 65.8\text{gMg} = \text{l. atoms Mg}$ Solve each of the following problems as directed. Show all your work. 11. Calculate the number Of atoms in 0.40 mol of sulfur- 12. Calculate the number of atoms in 2.30 mol of silver.

10 1 Review And Reinforcement Chemical Measurements Answer Key

11—2 Review and Reinforcement Solving Stoichiometry

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Problems If the statement is true, write "true." If it is false, change the underlined word or words to make it true. Write your answer on the line provided. 1. 2. 5. The major types of stoichiometry problems are mass-mass, mass-volume, and volume-volume.

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Chapter 9: Learning: Principles and Applications ...

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Reinforcement and Study Guide - Student Edition

24—1 Review and Reinforcement (continued) 13. There are 3.29 g of iodine-126 remaining in a sample originally containing 26.3 g of iodine-126. The half-life of iodine-126 is 13 days. How old is the sample? 14. A sample that originally contained 2.5 g of rubidium-87 now contains 1.25 g. The half-life of rubidium-87 is 6×10^{10} years.

KM 754e-20150121095452

Name 7—2 Review and Reinforcement Covalent Bonding C ass If

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the statement is true, write "true." If it is false, change the underlined word r words to make it

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